M. Sc. Degree Program in Chemistry

Program outcome

- To understand the advanced facts and concepts in chemistry
- To apply the principles of chemistry
- To develop research aptitude in chemistry
- To be able to work in pure, inter-disciplinary and multi-disciplinary areas of chemistry
- To familiarize with emerging areas of chemistry
- To develop the skills in the proper handling of the instruments and chemicals
- To Familiarize with the different processes used in industries and their applications
- To develop an ecofriendly attitude by creating a sense of environmental awareness

Course outcome

COURSE OUTCOME:

| Sl No | Name of the paper | Course Code | Course outcome |
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| 1 | QUANTUM MECHANICS AND COMPUTATION AL CHEMISTRY | CHE1C01 | To introduce the concept of Quantum mechanics among students To learn the theory and application of Quantum Chemistry Analyze the various possibilities of the concept in future research To learn different Quantum mechanical models, co-ordinate systems, wavefunctions etc To learn the basic principles of group theory and molecular symmetry To learn the method of representation of a point group and reduction formula To analyse the relation between quantum mechanics and group Theory |
| 2 | ELEMENTARY INORGANIC CHEMISTRY | CHE1C02 | To distinguish between different acid base concepts To analyse the molecular structure and bonding in molecules To learn the preparation, reactions and bonding in Boron hydrides To understand the different allotropic forms of C,S,P,As,Sb,Bi etc |

| STRUCTURE, AND REACTIVITY OF ORGANIC COMPOUNDS | CHE1C03 | To study the synthesis structure and uses of various compounds of main group elements To familiarize with the structure of nucleus, fission, fusion reactions etc To understands the interaction of radiation with matter To understand the basic principles of structure and bonding in organic molecules To learn the importance of different substituent effects and their application in organic chemistry To study the conformational analysis of various organic molecules an their application To learn the effect of conformation on the course of reaction rate in different system of molecules To understand the basic concepts of stereo chemisty To introduce the concept of Asymmetric synthesis and its application in various fields especially in research. |
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| THERMODYNA MICS, KINETICS AND CATALYSIS | CHE1C04 | To learn the basic theories and equations in Thermodynamics. To study the kinetic aspects of chemical reactions via molecular reaction dynamics To study thermodynamics of ideal and non ideal solutions To study homogeneous and heterogeneous catalysis |
| GROUP THEORY and CHEMICAL BONDING | CHE2C05 | To learn the different approximation methods in Quantum mechanics To study the Quantum mechanics of many electrom atoms To apply the theories of quantum mechanics in bonding of diatomic and polyatomic molecules To learn molecular vibrations of molecules and apply group theory in noleculer spectroscopy To apply group theory in chemical bonding |
| COORDINATIO N CHEMISTRY | CHE2C06 | To analyse the stability of coordination complexes To learn theories of bonding in coordination |

| | | To familiarise with electronic and magnetic properties of complexes To study characterisation techniques in coordination complexes To study the mechanisms of reactions of met complexes in detail To analyse the mechanisms of redox and photochemical reactions of complexes |
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| REACTION MECHANISM IN ORGANIC CHEMISTRY | CHE2C07 | To gain knowledge of the mechanisms of aliphatic and aromatic nucleophillic and electrophillic substitution reactions. To learn addition and elimination reactions and reactive intermediates To understand the chemistry of carbonyl compounds To study the pericyclic reactions in detail To learn photochemistry of organic compounds To understand the chemistry of natural products |
| ELECTROCHE MISTRY, SOLID STATE CHEMISTRY AND STATISTICAL THERMODYNA MICS | CHE2C08 | To study the dynamic electrochemistry To learn ionic interactions and equilibrium electrochemistry To study the crystal structure, imperfections and electronic structure of solids To learn quantum statistics of statistic thermodynamics |
| INORGANIC CHEMISTRY PRACTICALS-I & II | CHE1L01 & CHE2L04 | To study the seperation and identification of four met cations of which one is rare metal ion. To learn to carry our the spot tests of elements effectively To familiarise cerimetric and colourimetric estimation of metals |
| ORGANIC CHEMISTRY PRACTICALS– I & II | CHE1L02 & CHE2L05 | To learn the methods of Separation and Purification of Organic Compounds Analysis of binary mixtures, some of which containing compounds with more than one functional group To carry out multiple stage organic preparation |
| PHYSICAL CHEMISTRY | CHE1L03 & CHE2L06 | • To carry out different physical experiments with precision and accuracy |

| PRACTICALS – I &II MOLECULAR SPECTROSCOP Y | CHE3C09 | To plot graph efficiently and find out the unknown measurement from the graph To familiarise with electronic analytical instruments like potentiometer and conductometer To learn basic aspects of microwave, Infrared, NMR and Electronic spectroscopy To study the applications of Electronic and vibrational spectroscopy in Organic chemistry To familarise with the proton and C-13 NMR analysis in different organic molecules To learn Mass spectroscopy and its application in structural elucidation of organic compounds |
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| ORGANOMETA LLIC AND BIOINORGANI C CHEMISTRY | CHE3C10 | To learn the historical background and Nomenclature of organometallic compounds To study the organometallic compounds with linear and cyclic pi system To get knowledge about different reactions of organometallic compounds and catalysis To study about metal clusters and carbonyl clusters To learn the importance of bio-inorganic chemistry in various phases of biological system |
| REAGENTS AND TRANSFORMA TIONS IN ORGANIC CHEMISTRY | CHE3C11 | To study in detail Oxidation and reduction reactions of organic compounds To learn various synthetic reagents and their role in different reactions To learn classification of polymers and importance of Bio polymers and chemistry of hetero cyclic compounds To study various molecular rearrangements and transformations |
| SYNTHETIC ORGANIC CHEMISTRY (ELECTIVE) | CHE3E01 | To learn the reagents used for oxidation and reduction reactions To familiarize with various organometallic and organonon-metallic reagents To introduce different coupling reactions in organic chemistry To introduce the concept of Retro-synthetic analysis and multi-step synthesis |

| INSTRUMENTA L METHODS OF ANALYSIS INDUSTRIAL CATALYSIS (ELECTIVE) | CHE4C12 CHE4E05 | To learn about the errors in chemical analysis To study in detail about the conventional analytical methods To learn electro Analytical, optical, Thermal and Radio-chemical methods To study the theory and applications of chromatography To get an introduction to adsorption process To learn the preparative methods of a catalyst To study the deactivation of catalyst |
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| NATURAL PRODUCTS AND POLYMERS (ELECTIVE) | CHE4E05 | To introduce phase-transfer catalysis and Biocatalysts To study in depth about the natural products Brief introduction to dyes and pigments Introduction to polymers and polymerization processes Applications of polymers |
| INORGANIC CHEMISTRY PRACTICALS– III & IV | CHE3L07 & CHE4L10 | To develop analytical skills in inorganic quantitative analysis To understand ion exchange separation and estimation of binary mixture To understand the principles behind the colorimetric and to apply it in quantitative analysis To prepare inorganic complexes |
| ORGANIC CHEMISTRY PRACTICALS– III & IV | CHE3L08 & CHE4L11 | To develop talent in organic preparations to ensure maximum yield To apply the concept of melting or boiling pints to check the purity of compounds To analyze and characterize symbol organic functional group To analyze individual amino acids from a mixture using chromatography |
| PHYSICAL CHEMISTRY PRACTICALS– III & IV | CHE3L09 & CHE4L12 | To develop analytical skills in determining the physical properties To understand the principles of refractomery , potentiometry and conductometry |

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